









formed over a cylindrical core. The multilayers are each comprised of sublayers in order as follows: a copper sublayer with "nano-chicken wire" embedded in the copper sublayer for current collection, a nanostructured aluminum substrate sublayer, a nanostructured cathode sublayer, an electrolyte sublayer, a nanostructured anode sublayer, and a copper interlayer sublayer. The multilayers are repeated two or more times on the cylindrical core to form embodiments of the Nanostructure Lithium Ion Battery.

The nanostructured aluminum substrate sublayer, nanostructured cathode sublayer, and the nanostructured anode sublayer, are formed from an oblique sputtering process. When the material in question (the aluminum, cathode or anode material) are to be formed the sputtering head is positioned such that the normal to the sputtering target is at an oblique angle to the surface of the core sublayer. The oblique angle can range between 45 and 60 degrees from the normal to the surface of the core sublayer. The sputtering and the oblique angle forms nanostructures called "nanobaskets" that dramatically increase the surface area of the material in question leading to enhanced performance of embodiments.

In embodiments of the of the Nanostructure Lithium Ion Battery, the nanostructured aluminum substrate sublayer, nanostructured cathode sublayer, and the nanostructured anode sublayer, are comprised of nanobaskets formed from the oblique sputtering process.

A nanostructured electrode comprised of numerous "nanobaskets" provides a framework for enhanced use of the materials involved, at a fraction of the physical size of the battery. The layer thicknesses of the nanocomposite battery are not on the order of millimeters (such as a typical coin cell battery), but rather, in the vicinity of hundreds of nanometers to several microns. The battery has shown to be functional at an active region thickness (not counting current collectors) of less than 13 microns, and these thin layer structures can be manufactured in layers large enough to meet the end application.

The manufacturing processes involved in building a corndog battery is the use of oblique radio-frequency ("RF") sputtering. RF Sputtering of oxides onto smooth surfaces at oblique angles produces periodic surface corrugations. FIG. 8 illustrates the concept.

The advantage of this technique in the context of the corndog cell is that when sputtering a cylindrical sample, oblique incidence is by default the natural orientation of the incident particle's path. So, if the cylindrical core can be rotated inside the sputtering system at a pre-determined rate, the formation of a porous nanostructured electrode can be achieved quickly, with no need for a pre-formed template. The primary difference between the surface features formed in an oblique system versus a normally-incident system is that the surface features are positive rather than negative. That is, rather than forming nanobaskets on top of a defined structure of pits, the pores are essentially "nanohills" grown on top of an originally smooth metallic film.

The length of a thin film battery cell of this type could be made large, while the layer thickness remains on the order of microns. This stacked nanocomposite battery would have hundreds of thin film cells inside it, where a typical battery only has a handful.

The nanocomposite battery is mechanically flexible, owing to the presence of the polymer electrolyte. The polymer electrolyte's gel-like texture soaks into the structured electrode's pores, and provides a medium for lithium ion mobility as well as physical stability of the structure. The benefits of the corndog structure are numerous, not the least of which is the potential to manufacture high-performance batteries into flexible materials such as textiles or garments.

The geometry of the corndog cell involves a single, cylindrical thin film battery cell cascaded in series of four concentric rings. Each four-cell ring is then stacked concentrically. The stacking of four-cell rings is repeated indefinitely, until the entire battery cell stack reaches its desired thickness. The basic premise is shown below in FIGS. 4 and 5.

First, a 22.3 micron-thick copper wire is secured into place around a cylindrical core by sputtering a thin copper film over the core and wire. The thin copper layer acts as a current collector for the nanostructured anode layer, which is RF sputtered on top of the copper layer. The inner stackup is rotated about the z-axis in order to ensure even coating. The next layer outwards is the PEO electrolyte layer, in which ceramic microspheres are present. The PEO/microsphere layer is several microns thick, and is either sputtered or

"painted" on using an aerosol-like process. On top of this layer is sputtered the cathode, which may be either nanostructured or solid. If the cathode is nanostructured it is formed in the same fashion as the anode. Next, a thin layer of copper is sputtered onto the outer layer to act as a current collector.

Another winding of thin film battery cell is sputtered on top of the stack, followed by another copper current collector layer, and so forth, until a total of four battery cells have been laid down on top of each other. On top of the outer copper current collector layer, a new length of copper wire is added to the outside edge of the four-cell stack, followed by a battery separator material such as a porous polymer battery separator membrane manufactured from polyethylene or polypropylene. A porous polymer battery separator membrane impregnated with non-lithiated PEO works well as both an electrical insulator and a structural stabilizer.

Calculations show that the corndog battery achieves significantly greater energy densities over the Energizer CR2032. It is predicted that the energy density of the corndog cell is over 5 times the Energizer feature by weight and over 7 times the energy density of the Energizer by volume.

A nanocomposite battery is formed by plasma vapor deposition of constituent layers of a multiple layer assembly. The method of forming a nanocomposite battery comprises the steps of a) sputtering a conductive layer onto a substrate; b) sputtering an aluminum layer at an oblique angle onto the surface of the conducting layer; c) sputtering an anode material layer on the surface of the aluminum; d) sputtering an electrolyte layer onto the anode layer; e) sputtering a cathode material layer on the surface of the electrolyte; f) sputtering a conducting layer on the surface of the cathode; g) repeating of steps a through f until the required number of layers are formed, and h) sealing the layers by sputtering insulating polymer strips.

The method of determining the alternating current electrical characteristics of a coating is comprised of the steps of a) coating the inner conductor of an air coaxial transmission line with the coating to be analyzed, b) performing an impedance analysis of the coating and coaxial line to determine the physical and electrical properties of the coating as an electrolyte for a nanocomposite battery system of multilayered nanostructured materials.

The method of forming a nanocomposite battery comprising the isolation of the end portions of the multilayered nanocomposite battery. The method of forming a current collecting structure wherein anode connections emanate from one end of the battery and the cathode connections at the other end. At each end a method of integrating the connections to make external battery connections and mechanical support are employed.

The method of forming a nanocomposite battery comprises the use of oblique sputtering to form a nanoscale electrode template.

The method of forming a nanocomposite battery comprises the steps of adding layers of mechanical enhancers to the multilayered assembly for the purposes of strengthening the mechanical properties of the overall multilayered battery assembly.

A carrier for the multilayered nanocomposite battery to determine the mechanical properties of the nanocomposite battery comprises a) a platform with Kelvin probe compliant contact material; b) a piston structure; c) a compliant electrical contact using Kelvin probe; d) an adjustable spring force assembly; e) an atmospheric seal; and an external heater interface. The cell carrier is comprised of a "mesa" type architecture, and allows variable pressure to be applied to the battery terminals. The cell carrier also allows easy access to the battery terminals. The cell carrier is compatible with operation inside a glove box environment, and its rugged mechanical integrity allows precise measurements to be made in a repeatable manner.

The method of analyzing the performance of the multilayered nanocomposite battery comprises: a) a prototype cell carrier; b) a charge/discharge unit electrical connections; c) a cell loading and unloading means under controller atmosphere; and d) network analyzer connection for dynamic impedance analysis.

The method of forming a nanocomposite battery comprises the steps of sputtering nanocomposite or nanostructured electrodes using oblique sputtering to form the nanostructured electrodes. The method of forming a nanocomposite battery can be manufactured where in all the layers are formed by sputtering.

